Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

S. Thamotharan,<sup>a</sup>‡ V. Parthasarathi,<sup>a</sup>\* P. Sathishkumar,<sup>a</sup> Prashant R. Latthe,<sup>b</sup> Bharati Badami<sup>b</sup> and K. Ravikumar<sup>c</sup>

<sup>a</sup>Department of Physics, Bharathidasan University, Tiruchirappalli 620 024, India, <sup>b</sup>Post-Graduate Department of Studies in Chemistry, Karnatak University, Dharwad 580 003, India, and <sup>c</sup>Laboratory of X-ray Crystallography, Indian Institute of Chemical Technology, Hyderabad 500 007, India

Present address: Molecular Biophysics Unit, Indian Institute of Science, Bangalore 560 012, India.

Correspondence e-mail: vpsarati@yahoo.com

#### Key indicators

Single-crystal X-ray study T = 273 KMean  $\sigma$ (C–C) = 0.003 Å R factor = 0.043 wR factor = 0.116 Data-to-parameter ratio = 11.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

 $\odot$  2004 International Union of Crystallography Printed in Great Britain – all rights reserved

# *N*-(4-Acetylphenyl)-*N*'-[(4-sydnon-3-yl)phenyl]urea dimethylformamide solvate

In the title compound,  $C_{17}H_{14}N_4O_4 \cdot C_3H_7NO$ , the dihedral angle between the planes of the sydnone ring and the attached benzene ring is 13.56 (8)°. The benzene rings are inclined at an angle of 21.02 (8)° with respect to one another. In the solid state, intermolecular  $N-H \cdot \cdot \cdot O$  hydrogen bonds and weak intermolecular  $C-H \cdot \cdot \cdot O$  interactions are observed. Received 4 May 2004 Accepted 7 May 2004 Online 15 May 2004

### Comment

Sydnones are less toxic, but potent, porphyrinogenic and antiinflammatory compounds (Thamotharan, Parthasarathi, Sanyal *et al.*, 2003; Thamotharan, Parthasarathi, Hunnur *et al.*, 2003, and references therein). The present study was undertaken to determine the crystal and molecular structure of the title compound, (I) (Fig. 1).



The bond lengths and angles in the sydnone moiety in (I) are comparable to those of related compounds (Ueng *et al.*, 1985; Thamotharan, Parthasarathi, Sanyal *et al.*, 2003; Thamotharan *et al.*, 2004). The dihedral angle between the sydnone moiety and the attached benzene ring in 3-substituted sydnone derivatives ranges from 2 to  $39^{\circ}$  (Ueng *et al.*, 1987). In (I), the dihedral angle between the planes of the sydnone ring and the attached benzene ring is  $13.56 (8)^{\circ}$ . The two benzene rings are inclined at an angle of  $21.02 (8)^{\circ}$  with respect to one another.

In the crystalline state, carbonyl atom O1A of dimethylformamide accepts two different intermolecular N-H···O hydrogen bonds (Table 1) from the same neighbouring molecule, one from the N12-H12 group and the other from the N14-H14 group. These hydrogen bonds lead to an  $R_2^1(6)$ 



Figure 1

View of the molecule of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level.

motif (Bernstein et al., 1995). Atom C4 acts as a donor for a weak intermolecular C-H···O interaction with atom O13 of an adjacent molecule. This interaction links the molecules into a chain, which runs parallel to the b axis and has a graph-set motif of C(10). Atom C2A of the DMF molecule has a weak intermolecular C-H...O interaction with atom O5 of the sydnone moiety of an adjacent molecule. An intramolecular short contact is observed between atoms H12 and H14 (1.99 Å).

## Experimental

The title compound was prepared by refluxing 3-(p-azidocarbonyl)phenylsydnone (0.001 M) with p-aminoacetophenone (0.001 M) in anhydrous toluene (15 ml) for 4 h at 393 K. The crystalline product that separated out was collected and washed with toluene and petroleum ether. The title compound, (I), was recrystallized from an ethanol-DMF mixture (m.p. 489-490 K).

#### Crystal data

$C_{17}H_{14}N_4O_4 \cdot C_3H_7NO$	Z = 2		
$M_r = 411.42$	$D_x = 1.387 \text{ Mg m}^{-3}$		
Triclinic, P1	Mo $K\alpha$ radiation		
a = 7.7954 (8) Å	Cell parameters from 1983		
b = 8.3922 (8) Å	reflections		
c = 16.9225 (17)  Å	$\theta = 2.0-25.0^{\circ}$		
$\alpha = 79.057 \ (2)^{\circ}$	$\mu = 0.10 \text{ mm}^{-1}$		
$\beta = 84.267 \ (2)^{\circ}$	T = 273 (2)  K		
$\gamma = 65.049 \ (2)^{\circ}$	Prism, colourless		
V = 985.26 (17) Å <sup>3</sup>	$0.30 \times 0.22 \times 0.14 \text{ mm}$		
Data collection			
Siemens SMART CCD area-	2493 reflections with $I > 2\sigma(I)$		
detector diffractometer	$R_{\rm int} = 0.016$		
$\omega$ scans	$\theta_{\rm max} = 25.0^{\circ}$		
Absorption correction: none	$h = -8 \rightarrow 8$		
5038 measured reflections	$k = -9 \rightarrow 9$		

#### Refinement

Refinement on $F^2$
$R[F^2 > 2\sigma(F^2)] = 0.043$
$wR(F^2) = 0.116$
S = 1.03
3153 reflections
282 parameters
H atoms treated by a mixture of
independent and constrained
refinement

3153 independent reflections

 $l=-20\rightarrow 20$ 

$w = 1/[\sigma^2(F_o^2) + (0.053P)^2]$
+ 0.2508P]
where $P = (F_o^2 + 2F_c^2)/2$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.16 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.13 \text{ e } \text{\AA}^{-3}$

#### Table 1 Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
N12-H12···O1 $A^{i}$	0.85 (2)	2.07 (2)	2.880 (2)	159.3 (19)
$N14 - H14 \cdots O1A^{1}$	0.88(2)	1.99 (2)	2.831 (2)	159 (2)
$C4 - H4 \cdots O13^{n}$	0.93	2.44	3.332 (3)	161
$C2A - H2A \cdots O5^{m}$	0.93	2.36	3.267 (3)	166

Symmetry codes: (i) 1 - x, -y, 1 - z; (ii) x, y - 1, z; (iii) x, 1 + y, 1 + z.

The positions of the amine H atoms were determined from a difference Fourier map and refined freely, along with their isotropic displacement parameters. The methyl H atoms were constrained to an ideal geometry (C-H = 0.96 Å), with  $U_{iso}(H) = 1.5U_{eq}(C)$ . All remaining H atoms were placed in idealized positions (C-H = 0.93 Å) and constrained to ride on their parent atoms, with  $U_{iso}(H) =$  $1.2U_{eq}(C).$ 

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SIR92 (Altomare et al., 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2003).

#### References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). J. Appl. Cryst. 27, 435.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555-1573.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
- Thamotharan, S., Parthasarathi, V., Hunnur, R. K., Shinge, P. S., Badami, B. & Schenk, K. J. (2003). Acta Cryst. E59, 01762-01763.
- Thamotharan, S., Parthasarathi, V., Mallur, S., Kamble, R., Badami, B. & Linden, A. (2004). Acta Cryst. E60, o701-o702.
- Thamotharan, S., Parthasarathi, V., Sanyal, R., Badami, B. & Linden, A. (2003). Acta Cryst. E59, 0894-0896.
- Ueng, C.-H., Wang, Y. & Yeh, M.-Y. (1985). Acta Cryst. C41, 1776-1779.
- Ueng, C.-H., Wang, Y. & Yeh, M.-Y. (1987). Acta Cryst. C43, 1122-1125.